



Infosafe No™	1CH1A	Issue Date : April 2012	RE-ISSUED by CHEMSUPP
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Product Name : **BARIUM HYDROXIDE**

Classified as hazardous

1. Identification

GHS Product Identifier	BARIUM HYDROXIDE	
Company Name	CHEM-SUPPLY PTY LTD (ABN 19 008 264 211)	
Address	38 - 50 Bedford Street GILLMAN SA 5013 Australia	
Telephone/Fax Number	Tel: (08) 8440-2000 Fax: (08) 8440-2001	
Recommended use of the chemical and restrictions on use	In organic preparations; in barium salts; in the manufacture of oil and grease additives; in barium soaps and chemicals; in the refinishing of beet sugar and animal and vegetable oils; as an alkalizing agent in water softening; as a sulfate removal agent in the treatment of water and brine; in boiler scale removal; as a depilatory agent; as a catalyst in the manufacture of phenol-formaldehyde resins; as insecticide and fungicide; as a sulfate-controlling agent in ceramics; as a purifying agent for caustic soda; as a steel carbonizing agent; in glass; synthetic rubber vulcanization; as a corrosion inhibitor; in drilling fluids, lubricants; in analytical chemistry and laboratory reagent.	
Other Names	Name	Product Code
	BARIUM HYDROXIDE LR	BL033
	Barium hydrate	
	Caustic Baryta	
	Barium Hydroxide Octahydrate	
	Barium Oxide Hydrate Octahydrate	
Other Information	EMERGENCY CONTACT NUMBER: +61 08 8440 2000 Business hours: 8:30am to 5:00pm, Monday to Friday.	

Chem-Supply Pty Ltd does not warrant that this product is suitable for any use or purpose. The user must ascertain the suitability of the product before use or application intended purpose. Preliminary testing of the product before use or application is recommended. Any reliance or purported reliance upon Chem-Supply Pty Ltd with respect to any skill or judgement or advice in relation to the suitability of this product of any purpose is disclaimed. Except to the extent prohibited at law, any condition implied by any statute as to the merchantable quality of this product or fitness for any purpose is hereby excluded. This product is not sold by description. Where the provisions of Part V, Division 2 of the Trade Practices Act apply, the liability of Chem-Supply Pty Ltd is limited to the replacement of supply of equivalent goods or payment of the cost of replacing the goods or acquiring equivalent goods.

2. Hazard Identification

GHS classification of the substance/mixture	Acute Toxicity - Inhalation: Category 4 Acute Toxicity - Oral: Category 4 Skin Corrosion/Irritation: Category 1B
Signal Word (s)	DANGER
Hazard Statement (s)	H302 Harmful if swallowed. H314 Causes severe skin burns and eye damage. H332 Harmful if inhaled.
Pictogram (s)	Corrosion, Exclamation mark



Precautionary statement – Prevention	P260 Do not breathe dust/fume/gas/mist/vapours/spray. P264 Wash thoroughly after handling. P271 Use only outdoors or in a well-ventilated area.
Precautionary statement – Response	P280 Wear protective gloves/protective clothing/eye protection/face protection. P301+P330+P331 IF SWALLOWED: rinse mouth. Do NOT induce vomiting. P303+P361+P353 IF ON SKIN (or hair): Remove/Take off immediately all contaminated clothing. Rinse skin with water/shower. P304+P340 IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing. P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses,



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Precautionary statement – Storage
Other Information

if present and easy to do. Continue rinsing.
P310 Immediately call a POISON CENTER or doctor/physician.
P330 Rinse mouth.
P332+P313 If skin irritation occurs: Get medical advice/attention.
P362 Take off contaminated clothing and wash before reuse.
P405 Store locked up.
A systemic poison that competes with potassium in the nervous system.

3. Composition/information on ingredients

Chemical	Solid				
Characterization					
Ingredients	<u>Name</u>	<u>CAS</u>	<u>Proportion</u>	<u>Hazard Symbol</u>	<u>Risk Phrase</u>
	Barium hydroxide octahydrate	12230-71-6	98-100 %	Xn, C	R34, R20/21

4. First-aid measures

Inhalation	Remove from exposure, rest and keep warm. Blow nose to remove as much dust as possible. If breathing has stopped, apply artificial respiration. If breathing is difficult, give oxygen. Seek medical attention.
Ingestion	Rinse mouth thoroughly with water immediately. Give plenty of water to drink. Never give anything by mouth to an unconscious person. If swallowed, do NOT induce vomiting. Seek medical attention immediately.
Skin	Wash affected area thoroughly with soap and water. Remove contaminated clothing and wash before reuse or discard. If symptoms develop seek medical attention.
Eye contact	Immediately irrigate with copious quantity of water for at least 15 minutes. Eyelids to be held open. Seek immediate medical assistance.
First Aid Facilities	Eye wash fountains and safety showers should be available for emergency use.
Advice to Doctor	Treat symptomatically and supportively. The use of d-Penicillamine as a chelating agent should be determined by qualified medical personnel. Sodium sulfate can be given in case of ingestion to precipitate out the barium as barium sulfate.
Other Information	For advice, contact a Poisons Information Centre (Phone eg Australia 13 1126; New Zealand 0800 764 766) or a doctor.

5. Fire-fighting measures

Hazards from Combustion	Barium oxide.
Products	
Specific Methods	Small fire: Use dry chemical, CO2 or water spray. If safe to do so, move undamaged containers from fire area. Large fire: Use dry chemical, CO2, foam or water spray - Do not use water jets. Cool containers with flooding quantities of water until well after fire is out. Avoid getting water inside containers.
Specific hazards arising from the chemical	Material does not burn. Fire or heat will produce irritating, poisonous and/or corrosive gases. Containers may explode when heated. Some may ignite combustibles (wood, paper, clothing, etc.) Contact with metals may evolve flammable hydrogen gas.
Hazchem Code	2X
Decomposition Temp.	Decomposes at high temperatures, resulting in toxic and corrosive products. Loses its water molecules between 100 - 780 °C.
Precautions in connection with Fire	Wear SCBA and chemical splash suit. Fully-encapsulating, gas-tight suits should be worn for maximum protection. Structural firefighter's uniform is NOT effective for these materials.

6. Accidental release measures

Spills & Disposal	ELIMINATE all ignition sources (no smoking, flares, sparks or flames) within at least 50m. Do not touch or walk through spilled material. Do not touch damaged containers or spilled material unless wearing appropriate protective clothing. Stop leak if safe to do so - Prevent entry into waterways, drains or confined areas. Cover with DRY earth, sand or other non-combustible material followed by plastic sheet to minimize spreading or contact with rain. DO NOT GET WATER INSIDE CONTAINERS.
Personal Protection	Wear protective clothing specified for normal operations (see Section 8)

7. Handling and storage



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Precautions for Safe Handling	Avoid ingestion and inhalation of dust. Avoid contact with eyes, skin and clothing. Avoid prolonged or repeated exposure. Minimize dust generation and accumulation. Keep container closed. Use only with adequate ventilation. In case of insufficient ventilation, wear suitable respiratory equipment. If you feel unwell, seek medical attention and show the label when possible. Wear suitable protective clothing. Under no circumstances eat, drink or smoke while handling this material. It is essential that all who come into contact with this material maintain high standards of personal hygiene ie. washing hands prior to eating, drinking or going to the toilet. Wash thoroughly after handling. Remove contaminated clothing and wash before reuse. Protect against physical damage. Isolate from incompatible substances. Containers of this material may be hazardous when empty since they retain product residues (dust, solids); observe all warnings and precautions listed for the product.
Conditions for safe storage, including any incompatibilities	Corrosives/poisons area. Store in tightly closed, airtight containers, in a cool, dry, well-ventilated area away from incompatible substances. Keep away from strong acids. Keep away from heat, sources of ignition and open flames. Protect against physical damage. Store protected from moisture and direct sunlight. Absorbs carbon dioxide from air. Air sensitive. Containers of this material may be hazardous when empty since they retain product residues (dust, solids); observe all warnings and precautions listed for the product.
Corrosiveness	Corrosive to metals such as aluminium and zinc.
Storage Regulations	Refer Australian Standard AS/NZS 4452:1997 'The storage and handling of toxic substances'.
Storage Temperatures	Store at room temperature (15 to 25 °C recommended).
Unsuitable Materials	Light metals.

8. Exposure controls/personal protection

Occupational exposure limit values	Name	STEL		TWA		Footnote
		mg/m3	ppm	mg/m3	ppm	
	Barium hydroxide octahydrate			0.5		Barium, soluble compounds (as Ba)
Other Exposure Information	A time weighted average (TWA) has been established for Barium, soluble compounds (as Ba) (Worksafe Aust) of 0.5 mg/m3. The exposure value at the TWA is the average airborne concentration of a particular substance when calculated over a normal 8 hour working day for a 5 day working week.					
Appropriate engineering controls	In industrial situations maintain the concentrations values below the TWA. This may be achieved by process modification, use of local exhaust ventilation, capturing substances at the source, or other methods.					
Respiratory Protection	Where ventilation is not adequate, respiratory protection may be required. Avoid breathing dust, vapours or mists. Respiratory protection should comply with AS 1716 - Respiratory Protective Devices and be selected in accordance with AS 1715 - Selection, Use and Maintenance of Respiratory Protective Devices. Filter capacity and respirator type depends on exposure levels. In event of emergency or planned entry into unknown concentrations a positive pressure, full-facepiece SCBA should be used. If respiratory protection is required, institute a complete respiratory protection program including selection, fit testing, training, maintenance and inspection.					
Eye Protection	The use of a face shield, chemical goggles or safety glasses with side shield protection as appropriate. Must comply with Australian Standards AS 1337 and be selected and used in accordance with AS 1336.					
Hand Protection	Hand protection should comply with AS 2161, Occupational protective gloves - Selection, use and maintenance. Recommendation: Excellent: NR latex, vinyl, nitrile, neoprene gloves.					
Footwear	Safety boots in industrial situations is advisory, foot protection should comply with AS 2210, Occupational protective footwear - Guide to selection, care and use.					
Body Protection	Clean clothing or protective clothing should be worn, preferably with an apron. Clothing for protection against chemicals should comply with AS 3765 Clothing for Protection Against Hazardous Chemicals.					
Hygiene Measures	Always wash hands before smoking, eating or using the toilet. Wash contaminated clothing and other protective equipment before storing or re-using.					

9. Physical and chemical properties

Form	Solid
Appearance	White powder.



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Odour	Odourless.
Decomposition Temperature	Decomposes at high temperatures, resulting in toxic and corrosive products. Loses its water molecules between 100 - 780 °C.
Melting Point	78 °C
Boiling Point	780 °C
Solubility in Water	Partially soluble in cold water (16.5 g/l, 0 °C; 56 g/l, 15 °C; 37.4 g/l, 20 °C; 76 g/l, 40 °C; 173.2 g/l, 60 °C; 503.5 g/l, 80 °C).
Solubility in Organic Solvents	Readily soluble in dilute acids; slightly soluble in water, methanol, ethanol; insoluble in acetone.
Specific Gravity	2.18 (@ 16 °C)
pH	Aqueous solutions are strongly alkaline; 12.5 (50 g/l, H ₂ O, 20 °C).
Vapour Pressure	0 mm Hg; ca. 300 hPa at 77.9 °C.
Evaporation Rate	Negligible.
Volatile Component	0 %vol @ 21 °C
Flammability	Non combustible material.
Molecular Weight	315.47

10. Stability and reactivity

Chemical Stability	Stable under ordinary conditions of use and storage. Rapidly absorbs carbon dioxide from air to form nonvolatile carbonate, becoming incompletely soluble in water.
Conditions to Avoid	Air, excess heat, dust generation and incompatibles.
Incompatible Materials	Strong oxidizing agents, acids, metals such as aluminium and zinc, chlorinated rubber, fuels, reducing agents, acid halides and hydrogen sulfide.
Hazardous Decomposition Products	Barium oxide.
Possibility of hazardous reactions	Reaction with chlorinated rubber, with or without hydrocarbon or chlorinated solvents, is violent or explosive when heated at about 216 °C. Reactive with reducing agents, acids. Reacts with fuels.
Hazardous Polymerization	Has not been reported.

11. Toxicological Information

Acute Toxicity - Oral	LD50 (rat) = 550 mg/kg.
Ingestion	Harmful if swallowed. May cause severe irritation of the gastrointestinal tract, severe gastroenteritis, severe and permanent damage to the mouth, throat, oesophagus, and digestive tract, burns in oesophagus, gastrointestinal tract and stomach, risk of perforation in the oesophagus and stomach, tightness in the muscles of the face and neck, headache, nausea, vomiting, diarrhoea, abdominal pain, muscular tremors, dizziness, faintness, anxiety, weakness, labored breathing, paralysis of the arms and legs, cardiac dysrhythmia, increased blood pressure, CNS disorders, convulsions, collapse and death from cardiac and respiratory failure. May cause kidney failure. Estimated lethal dose lies between 1 to 15 grams. Death may occur within hours or up to a few days.
Inhalation	Harmful by inhalation. Inhalation of product vapours/dust will cause irritation of the nose, throat and respiratory system. Causes chemical burns to the respiratory tract. Symptoms include sore throat, coughing, wheezing, laryngitis, labored breathing, shortness of breath, headache, nausea and vomiting. Inhalation may result in respiratory effects such as inflammation, oedema, and chemical pneumonitis. Systemic poisoning may occur in sensitive individuals with symptoms similar to those of ingestion.
Skin	May cause severe irritation or burns in contact with skin, with possible redness, itching, scaling, or, occasionally, blistering. Solutions are strongly alkaline, highly irritating and may cause burns. May be harmful if absorbed through the skin.
Eye	Dusts cause eye irritation. On eye contact this product will cause tearing, stinging, blurred vision, and redness. Solutions may cause severe burns and damage. Eye contact may result in permanent damage and complete vision loss. Risk of blindness!
Carcinogenicity	Not listed in the IARC Monographs.
Chronic Effects	Chronic inhalation and ingestion may cause effects similar to those of acute exposure - severe and



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Serious eye damage/irritation	permanent damage to the digestive tract, respiratory tract and gastrointestinal tract burns, kidney failure, convulsions, increased blood pressure, muscle spasms, and possible paralysis. Prolonged and repeated exposure may cause damage to the blood, heart, cardiovascular system, bone marrow, spleen, the central nervous system, neuromuscular systems, kidneys, liver, mucous membranes, lungs and gastrointestinal system. Prolonged exposure may cause skin irritation.
Mutagenicity	Eye irritation test, rabbit: highly irritating.
Skin corrosion/irritation	No evidence of mutagenic properties.
	Skin irritation test, rat, rabbit: highly irritating.

12. Ecological information

Ecotoxicity	Quantitative data on the ecological effect of this product are not available. Toxic for aquatic organisms. Hazard for drinking water supplies. Harmful effect due to pH shift.
Environmental Protection	Do not allow to enter waters, waste water, or soil!
Acute Toxicity - Fish	The following applies to barium compounds: fish: lethal as from 158 mg/l: Salmo lethal as from 158 mg/l (as BaCl ₂); LC50 (L. idus): 870 mg/l (as BaCl ₂).
Acute Toxicity - Algae	The following applies to barium compounds: barium ions toxic for aquatic organisms: algae: Sc. quadricauda toxic as from 34 mg/l.
Acute Toxicity - Other Organisms	The following applies to barium compounds: crustaceans: toxic as from 29 mg/l.

13. Disposal considerations

Disposal Considerations	Whatever cannot be saved for recovery or recycling should be disposed of according to relevant local, state and federal government regulations.
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14. Transport information

Transport Information	Dangerous Goods of Class 8 Corrosives are incompatible in a placard load with any of the following: - Class 1, Class 4.3, Class 5, Class 6, if the Class 6 dangerous goods are cyanides and the Class 8 dangerous goods are acids and Class 7.
U.N. Number	2923
UN proper shipping name	CORROSIVE SOLID, TOXIC, N.O.S.
Transport hazard class(es)	8
Sub.Risk	6.1
Hazchem Code	2X
Packaging Method	3.8.8
Packing Group	II
EPG Number	8C3
IERG Number	37

15. Regulatory information

Poisons Schedule	S6
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16. Other Information

Literature References	'Standard for the Uniform Scheduling of Medicines and Poisons No. 6', Commonwealth of Australia, February 2015. Lewis, Richard J. Sr. 'Hawley's Condensed Chemical Dictionary 13th. Ed.', Rev., John Wiley and Sons, Inc., NY, 1997. National Road Transport Commission, 'Australian Code for the Transport of Dangerous Goods by Road and Rail 7th. Ed.', 2007. Safe Work Australia, 'National Code of Practice for the Preparation of Safety Data Sheets for Hazardous Chemicals', 2011. Standards Australia, 'SAA/SNZ HB 76:2010 Dangerous Goods - Initial Emergency Response Guide', Standards Australia/Standards New Zealand, 2010.
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Safety Data Sheet

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**Contact
Person/Point**

Safe Work Australia, 'Approved Criteria for Classifying Hazardous Substances [NOHSC:1008 (2004)]'.
Safe Work Australia, 'Hazardous Substances Information System, 2005'.
Safe Work Australia, 'National Code of Practice for the Labelling of Safe Work Hazardous Substances (2011)'.
Safe Work Australia, 'National Exposure Standards for Atmospheric Contaminants in the Occupational Environment [NOHSC:1003(1995)]'.

Paul McCarthy Ph. (08) 8440 2000 **DISCLAIMER STATEMENT:**

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**Empirical Formula &
Structural Formula**

Ba(OH)₂·8H₂O

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