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Infosafe No™ RE-ISSUED by CHEMSUPP 1CHIU Issue Date: September 2016

Product Name: SILVER NITRATE Solution

Classified as hazardous

1. Identification

**GHS Product** 

SILVER NITRATE Solution

Identifier

CHEM-SUPPLY PTY LTD (ABN 19 008 264 211) **Company Name** 

38 - 50 Bedford Street GILLMAN **Address** 

SA 5013 Australia

Telephone/Fax Tel: (08) 8440-2000 Fax: (08) 8440-2001 Number Laboratory reagent. Recommended use

of the chemical and restrictions on use

**Other Names Name Product Code** 

> SILVER NITRATE 0.1M Solution LR SL143

Other Information **EMERGENCY CONTACT NUMBER:** +61 08 8440 2000

Business hours: 8:30am to 5:00pm, Monday to Friday.

Chem-Supply Pty Ltd does not warrant that this product is suitable for any use or purpose. The user must ascertain the suitability of the product before use or application intended purpose. Preliminary testing of the product before use or application is recommended. Any reliance or purported reliance upon Chem-Supply Pty Ltd with respect to any skill or judgement or advice in relation to the suitability of this product of any purpose is disclaimed. Except to the extent prohibited at law, any condition implied by any statute as to the merchantable quality of this product or fitness for any purpose is hereby excluded. This product is not sold by description. Where the provisions of Part V, Division 2 of the Trade Practices Act apply, the liability of Chem-Supply Pty Ltd is limited to the replacement of supply of equivalent goods or payment of the cost of replacing the goods or acquiring equivalent goods.

### 2. Hazard Identification

GHS classification Hazardous to the Aquatic Environment - Acute Hazard: Category 1

Hazardous to the Aquatic Environment - Long-Term Hazard: Category 1 of the

Eye Damage/Irritation: Category 2A substance/mixture

Skin Corrosion/Irritation: Category 2

Signal Word (s) **WARNING** 

**Hazard Statement** H315 Causes skin irritation.

H319 Causes serious eye irritation.

H410 Very toxic to aquatic life with long lasting effects.

Pictogram (s) Exclamation mark, Environment





**Precautionary** P264 Wash thoroughly after handling.

P280 Wear protective gloves/protective clothing/eye protection/face protection. statement -

P273 Avoid release to the environment. Prevention

P302+P352 IF ON SKIN: Wash with plenty of soap and water. **Precautionary** P332+P313 If skin irritation occurs: Get medical advice/attention. statement -P362 Take off contaminated clothing and wash before reuse. Response

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses,

if present and easy to do. Continue rinsing.

P337+P313 If eye irritation persists: Get medical advice/attention.

**Precautionary** statement -Disposal

P501 Dispose of contents/container to an approved waste disposal plant.

#### 3. Composition/information on ingredients

Chemical Liquid

Characterization

Ingredients <u>Name</u> CAS **Proportion** 



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Water 7732-18-5 98.3 % Silver nitrate 7761-88-8 1.7 %

4. First-aid measures

Inhalation Remove to fresh air.

Ingestion Rinse mouth thoroughly with water immediately, repeat until all traces of product have been removed.

DO NOT INDUCE VOMITING. Seek medical advice if effects persist.

Wash affected areas with copious quantities of water. Remove contaminated clothing and wash before Skin

re-use. If rapid recovery does not occur, obtain medical attention

If contact with the eye(s) occurs, wash with copious amounts of water for approximately 15 minutes Eve contact

holding eyelid(s) open. Take care not to rinse contaminated water into the non-affected eye. Seek

medical attention.

Maintain eyewash fountain and safety shower in work area. **First Aid Facilities** 

Treat symptomatically based on judgement of doctor and individual reactions of the patient. **Advice to Doctor** 

Other Information For advice, contact a Poisons Information Centre (Phone eg Australia 13 1126; New Zealand 0800 764

766) or a doctor.

5. Fire-fighting measures

Hazards from Oxygen, toxic fumes, toxic oxides of nitrogen, nitrous gases, metallic silver, silver oxides.

Combustion **Products** 

**Specific Methods** No limitations to the type of extinguishing media. Use extinguishing media most appropriate for the

surrounding fire. Does not burn.

Specific hazards

arising from the

chemical **Hazchem Code** 

Precautions in Wear SCBA and structural firefighter's uniform.

connection with Fire

6. Accidental release measures

Personal Avoid substance contact.

2X

**Precautions** 

Personal Protection Wear protective clothing specified for normal operations (see Section 8)

Clean-up Methods -**Small Spillages** 

Absorb or contain liquid with sand, earth or spill control material. Shovel up using non sparking tools and place in a labelled, sealable container for subsequent safe disposal. Put leaking containers in a labelled

drum or overdrum.

**Environmental Precautions** 

Prevent from entering into drains, ditches, rivers or the sea.

Other Information

Most organisations using silver compounds collect all silver residues for subsequent recovery. Solutions

could be treated with a salt solution and the resulting silver chloride collected for subsequent recovery.

7. Handling and storage

Precautions for Safe Avoid contact with eyes, skin, and clothing. Avoid ingestion. Avoid prolonged or repeated exposure. Wear suitable protective clothing. Wash thoroughly after handling.

Handling

Conditions for safe storage, including

Store in a tightly closed container. Store in a cool, dry, well-ventilated area away from incompatible substances. Keep well closed and protected from direct sunlight and moisture. Light sensitive. Aqueous solutions are incompatible with alkali and alkaline earth metals and many reactive organic and inorganic

chemicals. incompatabilities

Storage Regulations Refer Australian Standard AS 3780-1994 'The storage and handling of corrosive substances'.

Store at room temperature (15 to 25 °C recommended). **Storage** 

**Temperatures** 

Unsuitable Materials Organic materials.

8. Exposure controls/personal protection

Occupational exposure limit

values

**STEL** TWA **Name** 



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mg/m3 mg/m3 <u>ppm</u> **Footnote** ppm Silver nitrate 0.01 Silver, soluble compoun ds (as Ag)

Other Exposure Information

A time weighted average (TWA) has been established for Silver, soluble compounds (as Ag) (Safe Work Australia) of 0.01 mg/m³ and for Silver, metal (Safe Work Australia) of 0.1 mg/m³. The exposure value at the TWA is the average airborne concentration of a particular substance when calculated over a normal

8 hour working day for a 5 day working week.

**Appropriate** 

In industrial situations maintain the concentrations values below the TWA. This may be achieved by engineering controls process modification, use of local exhaust ventilation, capturing substances at the source, or other

methods.

Respiratory Protection

Normally not required.

The use of a face shield, chemical goggles or safety glasses with side shield protection as appropriate. **Eve Protection** Must comply with Australian Standards AS 1337 and be selected and used in accordance with AS 1336.

**Hand Protection** Hand protection should comply with AS 2161, Occupational protective gloves - Selection, use and

maintenance. Recommendation: Excellent: NR latex. Good: Nitrile rubber gloves

**Personal Protective** 

Final choice of personal protective equipment will depend on individual circumstances and/or according

to risk assessments undertaken.

Equipment **Body Protection** 

**Hygiene Measures** 

Clean clothing or protective clothing should be worn, preferably with an apron. Clothing for protection

against chemicals should comply with AS 3765 Clothing for Protection Against Hazardous Chemicals. Always wash hands before smoking, eating or using the toilet. Wash contaminated clothing and other

protective equipment before storing or re-using.

9. Physical and chemical properties

Form Liquid

Clear, colourless liquid. **Appearance** 

Odour Odourless. >2 °C **Melting Point** ~ 100 °C **Boiling Point** Miscible. Solubility in Water **Specific Gravity** ~ 1.04

3.0 - 6.0 (25 °C) рΗ 14 mm Hg @ 20 °C **Vapour Pressure** 

**Vapour Density** 

(Air=1)

**Evaporation Rate** >1 (ether=1)

Volatile Component ca. 98 %vol @ 21 °C **Flammability** Non flammable. Molecular Weight 169.87 + aq

Oxidising Properties Dried product residue is a strong oxidizer.

10. Stability and reactivity

**Chemical Stability** Stable at room temperature in closed containers under normal storage and handling conditions. Light

sensitive. Darkens on exposure to light.

**Conditions to Avoid** Light, air, heat, high temperatures, evaporating to near dryness and incompatible materials.

Incompatible **Materials** 

Due to low concentration of silver nitrate - none, exception being with substances reactive to water.

**Hazardous** 

Toxic fumes, toxic oxides of nitrogen, nitrous gases, metallic silver, oxygen, silver oxides.

Decomposition **Products** 

**Hazardous** Will not occur.

**Polymerization** 



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11. Toxicological Information

Acute Toxicity - Oral LD50 (rat): >2000 mg/kg (calculated on the pure substance).

Ingestion Causes gastrointestinal irritation with nausea, vomiting and diarrhoea. Effects should be less severe

than through exposure to higher concentrations of silver nitrate which may include symptoms of pain and burning in mouth, blackening of the skin and mucous membranes, throat and abdomen, diarrhoea and

shock. A poison in higher concentrations.

Inhalation Causes respiratory tract irritation. May be absorbed into the body following inhalation with symptoms

paralleling those from ingestion exposure.

Skin Causes skin irritation. May cause skin discolouration. Effects are expected to be less severe than for

exposure to higher concentrations where symptoms include redness, pain and severe burns.

Causes severe eye irritation, redness and pain. Contact may cause eye damage, corneal scarring, Eye

permanent corneal opacification.

**Chronic Effects** Nitrites/nitrates may cause methaemoglobinaemia, which is characterized by chocolate-brown coloured

blood, headache, weakness, dizziness, breath shortness, cyanosis (bluish skin due to deficient

oxygenation of blood), rapid heart rate, unconsciousness and possible death.

Chronic inhalation or ingestion of silver salts may cause argyria characterized by a permanent blue-gray discolouration of the eyes, skin, mucous membranes, and internal organs. This malady results from the

accumulation of silver in the body.

Silver nitrate solution: no evidence of mutagenic properties. Mutagenicity

Silver nitrate: Laboratory experiements have shown mutagenic effects.

12. Ecological information

**Ecotoxicity** Harmful effect on aquatic organisms. May cause long-term adverse effects in the aquatic environment.

Silver ions toxic for aquatic organisms.

**Environmental** 

Do not allow to enter waters, waste water, or soil!

**Protection** 

Acute Toxicity - Fish Silver nitrate

Leuciscus idus: LC50: 0.029 mg/L/96h

**Acute Toxicity -**Silver nitrate

Semi static test LC50 (D. daphnia): 0.0069 - 0.0082 mg/L/48h. Daphnia

**Acute Toxicity -**Silver nitrate

**Bacteria** Ps. putida toxic from 0.006 mg/L/16h.

13. Disposal considerations

Whatever cannot be saved for recovery or recycling should be disposed of according to relevant local, Disposal Considerations

state and federal government regulations.

14. Transport information

Dangerous Goods of Class 8 Corrosives are incompatible in a placard load with any of the following: -**Transport** Information

Class 1, Class 4.3, Class 5, Class 6, if the Class 6 dangerous goods are cyanides and the Class 8

dangerous goods are alkalis and Class 7.

**U.N. Number** 

UN proper shipping ENVIRONMENTALLY HAZARDOUS SUBSTANCE, LIQUID, N.O.S. - (Silver Nitrate Solution)

name

**Transport hazard** 

class(es)

9

**Hazchem Code** 

2X **Packaging Method** 3.8.9 **Packing Group** Ш

**EPG Number** 9C1 **IERG Number** 47

15. Regulatory information

Regulatory Listed in the Australian Inventory of Chemical Substances (AICS).

Information

**Poisons Schedule** 

S6

SL143: (S6) Other Information



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SL618: Not scheduled

#### 16. Other Information

### Literature References

'Standard for the Uniform Scheduling of Medicines and Poisons No. 6', Commonwealth of Australia, February 2015.

Lewis, Richard J. Sr. 'Hawley's Condensed Chemical Dictionary 13th. Ed.', Rev., John Wiley and Sons,

Inc., NY, 1997. National Road Transport Commission, 'Australian Code for the Transport of Dangerous Goods by Road

and Rail 7th. Ed.', 2007. Safe Work Australia, 'National Code of Practice fot the Preparation of Safety Data Sheets for Hazardous

Chemicals', 2011. Standards Australia, 'SAA/SNZ HB 76:2010 Dangerous Goods - Initial Emergency Response Guide',

Standards Australia/Standards New Zealand, 2010. Safe Work Australia, 'Approved Criteria for Classifying Hazardous Substances [NOHSC:1008 (2004)]'.

Safe Work Australia, 'Hazardous Substances Information System, 2005'.

Safe Work Australia, 'National Code of Practice for the Labelling of Safe Work Hazardous Substances (2011)'.

Safe Work Australia, 'National Exposure Standards for Atmospheric Contaminants in the Occupational Environment [NOHSC:1003(1995) 3rd Edition]'.

### Contact Person/Point

Paul McCarthy Ph. (08) 8440 2000 DISCLAIMER STATEMENT:

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#### Empirical Formula & AgNO3 + aq Structural Formula

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