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Product Name TIN METAL (Foil, Granules and Powder)

Not classified as hazardous

1. Identification

GHS Product Identifier TIN METAL (Foil, Granules and Powder)

Company Name

CHEM-SUPPLY PTY LTD (ABN 19 008 264 211)

Address

50 Bedford Street GILLMAN SA 5013 Australia

Telephone/Fax Number Tel: (08) 8440-2000 Fax: (08) 8440-2001

Recommended use of the chemical and restrictions on use Tin plated steel containers for food preservation, tin alloys (brasses and bronzes, bell metal, Babbitt metal, die casting alloy, pewter, phosphor bronze, type, and White metal), soldering alloys for electrical/electronic and general industrial applications, specialized alloys such as dental amalgams, titanium alloys used in aircraft engineering, niobium-tin and indium-tin alloys used in superconducting cables and magnets and indium-tin oxide for metallic photonic crystals, low-melting alloys for fire control, organ pipes, tin alloys are important in the production of coatings by electroplating and hot tinning (the most important of these are tin-zinc, tin-nickel, tin-cobalt, and tin-copper), tin coatings (applied to most metal surfaces by electrodeposition, while in hot-dipping, molten tin wets and adheres readily to clean iron, steel, copper, and copper-base alloys), corrosion-resistant coatings (for lead or zinc and steel), cladding, tinned wire (all copper wire that is to be rubber covered), collapsible tubes, anodes for electron plating, manufacture of chemicals (tin salts), block tin (used to coat copper cooking utensils and lead sheet, or to line lead pipe for distilled water, beer, carbonated beverages, and some chemicals), powder metallurgy applications, exothermic welding, catalysts, colours, stabilizer, cast and wrought forms, window glass manufacture ('Pilkington process') and transportation

applications.

Other Names Name

Product Code

TIN METAL Foil 0.2mm LR TL017
TIN METAL Granules LR TL018

Stannum

Other Information

EMERGENCY CONTACT NUMBER: +61 08 8440 2000 Business hours: 8:30am to 5:00pm, Monday to Friday.

Chem-Supply Pty Ltd does not warrant that this product is suitable for any use or purpose. The user must ascertain the suitability of the product before use or application intended purpose. Preliminary testing of the product before use or application is recommended. Any reliance or purported reliance upon Chem-Supply Pty Ltd with respect to any skill or judgement or advice in relation to the suitability of this product of any purpose is disclaimed. Except to the extent prohibited at law, any condition implied by any statute as to the merchantable quality of this product or fitness for any purpose is hereby excluded. This product is not sold by description. Where the provisions of Part V, Division 2 of the Trade Practices Act apply, the liability of Chem-Supply Pty Ltd is limited to the replacement of supply of equivalent goods or payment of the cost of replacing the goods or acquiring equivalent

2. Hazard Identification

GHS classification of

Not classified as hazardous according to the criteria of Hazardous Substances Infromation System (HSIS), Safe Work Australia.

substance/mixture

Not classified as dangerous goods according to the Australian Dangerous Goods Code (ADG).

3. Composition/information on ingredients

Chemical

Solid

Characterization

Ingredients Nam

Name CAS Proportion Hazard Symbol Risk Phrase

Tin 7440-31-5 100 %

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ppm

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4. First-aid measures

Remove from exposure, rest and keep warm. If breathing has stopped, apply Inhalation

artificial respiration. If breathing is difficult, give oxygen. Seek medical

attention in severe cases, or if symptoms develop.

Rinse mouth thoroughly with water immediately. Give plenty of water to drink. Ingestion

Never give anything by mouth to an unconscious person. If swallowed, do NOT Seek medical attention in severe cases, or if large amounts induce vomiting.

ingested.

Wash affected area thoroughly with copious amounts of running water. Remove Skin

contaminated clothing and wash before reuse. Seek medical attention in severe

cases, or if irritation develops.

If contact with the eye(s) occurs, wash with copious amounts of water for Eve contact

approximately 15 minutes holding eyelid(s) open. Take care not to rinse contaminated water into the non-affected eye. Seek medical attention.

Maintain eye wash and normal washroom facilities. First Aid Facilities

Advice to Doctor Treat symptomatically and supportively.

No action shall be taken involving any personal risk or without suitable **Protection for First**

training. Aiders

5. Fire-fighting measures

Hazards from Irritating and/or highly toxic fumes and gases, tin/tin oxides.

Wear SCBA and structural firefighter's uniform.

Combustion **Products**

Use extinguishing media most appropriate for the surrounding fire. Specific Methods

Small fire: Use dry chemical, CO2, water spray or foam.

Precautions in

connection with Fire

6. Accidental release measures

Avoid breathing dust and contact with skin and eyes. Spills & Disposal

Personal Protection Wear protective clothing specified for normal operations (see Section 8)

Clean-up Methods -**Small Spillages**

Sweep up and place in a labelled container for subsequent safe disposal.

7. Handling and storage

Precautions for Safe Handling

Avoid ingestion and inhalation of dust/granules/foil. Avoid contact with eyes, skin, and clothing. Avoid prolonged or repeated exposure. Minimize dust generation and accumulation. Keep containers closed when not in use. Use in designated areas with adequate ventilation. Keep away from incompatibles such

as oxidizing agents, acids, alkalis.

Conditions for safe storage, including any incompatabilities

exposure limit values

Other Exposure

Store in tightly closed containers, in a cool, dry, well-ventilated area away

from incompatible substances. Separated from oxidising agents.

8. Exposure controls/personal protection

Occupational Name STEL TWA

mg/m3 mg/m3 ppm

Tin A time weighted average (TWA) has been established for Tin, metal (Worksafe

Aust) of 2 mg/m³. The exposure value at the TWA is the average airborne Information concentration of a particular substance when calculated over a normal 8 hour

working day for a 5 day working week.

In industrial situations maintain the concentrations values below the TWA. **Appropriate** engineering controls

This may be achieved by process modification, use of local exhaust

ventilation, capturing substances at the source, or other methods.

Respiratory Where ventilation is not adequate, respiratory protection may be required. Avoid breathing dust, vapours or mists. Respiratory protection should comply **Protection** with AS 1716 - Respiratory Protective Devices and be selected in accordance

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with AS 1715 - Selection, Use and Maintenance of Respiratory Protective Devices. Filter capacity and respirator type depends on exposure levels. event of emergency or planned entry into unknown concentrations a positive pressure, full-facepiece SCBA should be used. If respiratory protection is required, institute a complete respiratory protection program including

selection, fit testing, training, maintenance and inspection.

The use of a face shield, chemical goggles or safety glasses with side shield **Eye Protection**

protection as appropriate. Must comply with Australian Standards AS 1337 and

be selected and used in accordance with AS 1336.

Hand protection should comply with AS 2161, Occupational protective gloves -**Hand Protection**

Selection, use and maintenance.

Clean clothing or protective clothing should be worn. Clothing for protection **Body Protection**

against chemicals should comply with AS 3765 Clothing for Protection Against

Hazardous Chemicals.

Always wash hands before smoking, eating or using the toilet. Wash **Hygiene Measures**

contaminated clothing and other protective equipment before storing or

re-using.

9. Physical and chemical properties

Form

Almost silver-white to gray, lustrous, malleable metal; cubic (gray); **Appearance**

tetragonal (white); rhomboidal (tin brittle); granules, foil, or powder. At $-40~^{\circ}\text{C}$ crumbles to gray amorphous powder (gray tin); slowly changes back above 20 $^{\circ}\text{C}$ to white tin; brittle at 200 $^{\circ}\text{C}$.

Odourless. Odour 231.9 °C. **Melting Point**

2270 °C; 2507 °C; 2602 °C. **Boiling Point**

Insoluble in water. Solubility in Water

Solubility in Organic

Soluble in hydrochloric acid, sulfuric acid, aqua regia, alkali; slightly

soluble in dilute nitric acid. **Solvents** 7.265 (white); 5.769 (gray). **Specific Gravity**

1.3332 hPa at 1492 °C. Vapour Pressure

1.85 mPa.s (cP) @ 240 °C. Viscosity

0 %vol @ 21 °C **Volatile Component**

544 mN/m (@ 231.9 °C). **Surface Tension** Non combustible material. **Flammability**

Auto-Ignition

Dust Cloud: 630 °C; Dust Layer: 430 °C.

Temperature

Flammable Limits -> 99.99 % (powder).

Lower

Explosion Properties The finely divided dust may form flammable/explosive mixtures with air. It may present a dust explosion hazard in the presence of an ignition source. Minimum

explosible concentration: 0.19 g/l. Particle size and air concentration

determine reactivity.

Tin reacts violently or explosively with fused ammonium nitrate below 200 °C.

Contact of metallic tin with turpentine may cause fires and explosions.

Molecular Weight 118.69.

Tin has two allotropic forms at normal pressure; at -40 °C crumbles to gray **Other Information**

amorphous powder (gray tin; alpha); slowly changes back above 20 °C to white tin (beta); brittle @ 200 °C.

Transformation temp: (beta in equilibrium with alpha) 13.2 °C. Resistivity of white tin: 11.0 μ -Ohm cm @ 0 °C; 15.5 μ -Ohm cm @ 100 °C; 20.0 μ -Ohm cm @ 200 °C; 22.0 μ -Ohm cm @ mp (solid); 45.0 μ -Ohm cm @ mp (liquid).

10. Stability and reactivity

Chemical Stability Stable against air and water under normal temperatures, pressures and conditions of handling and storage. Powder oxidizes, especially in the

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presence of air and moisture. At high temperatures, it burns with an intensive white flame and forms tin oxide.

Conditions to Avoid

Excess heat, flames, ignition sources, dust generation and incompatible materials. Powder is air and moisture sensitive.

Incompatible Materials Oxidizing agents (sodium peroxide and potassium peroxide, potassium dioxide, fused ammonium nitrate below 200 °C, cupric nitrate, in the presence of water), strong acids (generation of hydrogen) (nitric acid, hydrochloric acid), strong bases, halogens and halogen trifluorides (fluorine at 100 °C, heat + chlorine, tin chloride, carbon tetrachloride, in the presence of water vapour, bromine, chlorine trifluoride in the presence of carbon, disulfur dichloride), sulfur, some extinguishing agents such as bicarbonate powder and carbon dioxide, tellurium, turpentine, water + heat, mixtures with air in the presence of an ignition source.

Hazardous
Decomposition
Products
Possibility of
hazardous reactions

Toxic and/or irritating fumes and gases, tin/tin oxides.

Reacts with strong oxidants. Tin reacts violently or explosively with fused ammonium nitrate below 200 $^{\circ}\text{C}$. In the presence of water, cupric nitrate and tin foil, on prolonged and intimate contact, will produce flaming and sparking. Sodium peroxide and potassium peroxide, potassium dioxide, oxidize $\ensuremath{\operatorname{tin}}$ with incandescence. Reacts violently with strong acids and some extinguishing agents such as bicarbonate powder and carbon dioxide. Reactive with alkalis. The violent reaction between tin and bromine is controlled in halocarbon solutions. Tin /begins to burn at 100 $^{\circ}\text{C}$ in fluorine. Reaction with chlorine trifluoride, in the presence of carbon, is violent. Reacts violently with bromine trifluoride. Tin reacts violently with Iodine Bromide. When heated in Chlorine, Tin reacts, producing light and much heat. Interaction with carbon tetrachloride, in the presence of water vapour, is violent. Interaction with disulfur dichloride is violent. Reaction with sulfur is vigorous and accompanied by incandescence. Contact of metallic tin with turpentine may cause fires and explosions. The reaction between tin and tellurium attains incandescence. Experiments involving explosions of molten tin and water are described. The finely divided dust may form flammable/explosive mixtures with air. It may present a dust explosion hazard in the presence of an ignition source, when exposed to heat or by spontaneous chemical reaction with Br2, BrF3, S, Cl2,ClF3, Cu(NO3), K2O2. Will not occur.

Hazardous Polymerization

11. Toxicological Information

Ingestion

May be harmful if swallowed. Ingested metallic tin exhibits only moderate toxicity due to poor absorption from the digestive tract and rapid tissue turnover. Ingestion of large doses of powdered tin may cause gastrointestinal irritation, nausea, cramps, vomiting, and diarrhoea (which may be from irritant or astringent action on the stomach), but not permanent injury. Inorganic tin salts, which may form with corrosion depending on a number of factors, including the presence of oxidising agents (oxygen, nitrate) and acids, may cause nausea, vomiting and diarrhoea, may interfere with various enzyme systems and may cause systemic effects on the central nervous system, heart and liver, if ingested in concentrations in excess of 300-500 mg/kg. Inhalation of tin dust may cause irritation, due to mechanical action, to nose, throat and respiratory tract, with coughing. Inhaled dust or fumes may cause benign, symptomless pneumoconiosis (stannosis). This form of pneumoconiosis produces distinctive progressive x-ray changes of the lung as long as exposure persists, but there is no distinctive fibrosis, no evidence of disability, and no special complicating factors. Inhalation of fumes may cause metal fume fever, which is characterized by flu-like symptoms with metallic taste, fever, chills, cough, weakness, chest pain, muscle pain and increased white blood cell count.

Inhalation

May cause mild skin irritation, resulting in redness and itchiness.

Skin Eye

Dust may cause eye irritation due to mechanical action, with redness and pain. Granules and foil may cause eye damage due to mechanical action.

Carcinogenicity

Not listed in the IARC Monographs.

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Reproductive **Toxicity**

Tin [resp/skin]: animal-possible increase in subtle neurological & skeletal deformities (from: 'Reproductive Hazards of the Workplace' by Linda M.

Frazier, MD, MPH & Marvin L. Hage, MD).

Prolonged and/or repeated contact may cause irritation and/or dermatitis. Chronic Effects

Chronic exposure to dust or fumes may have effects on the lungs, resulting in a benign pneumoconiosis (stannosis). This form of pneumoconiosis produces distinctive progressive x-ray changes of the lung as long as exposure

persists, but there is no distinctive fibrosis, no evidence of disability, and

no special complicating factors.

12. Ecological information

Ecological Information **Ecotoxicity**

No ecological problems are to be expected when the product is handled and used

with due care and attention.

Quantitative data on the ecological effect of this product are not available.

Environmental Protection

Do not allow to enter waters, waste water, or soil!

13. Disposal considerations

Disposal Considerations Dispose of according to relevant local, state and federal government

regulations.

14. Transport information

15. Regulatory information

Not Scheduled Poisons Schedule

16. Other Information

Literature References

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Australia.

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Combustible Liquids.

Standards Australia, 'SAA/SNZ HB 76:2010 Dangerous Goods - Initial Emergency

Response Guide', Standards Australia/Standards New Zealand, 2010.

Worksafe Australia, 'Approved Criteria for Classifying Hazardous Substances

[NOHSC:1008(2004)]'.

Worksafe Australia, 'Hazardous Substances Information System, 2005'.
Worksafe Australia, 'National Code of Practice for the Labelling of Workplace

Substances [NOHSC:2012(1994)]'.

Worksafe Australia, 'National Exposure Standards for Atmospheric Contaminants in the Occupational Environment [NOHSC:1003(1995)]'.

Contact Person/Point Paul McCarthy Ph. (08) 8440 2000 DISCLAIMER STATEMENT:

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Empirical Formula & Structural

Formula

User Codes User Field Title

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